COMPUTATIONAL SIMULATION OF FIRE DEVELOPMENT INSIDE A TRADE CENTRE

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#### Abstract

: Real scale fire experiments involve considerable costs compared to computational mathematical modelling. This paperwork is the result of such a virtual simulation of a fire occurred in a hypothetical wholesale warehouse comprising a large number of trade stands. The analysis starts from the ignition source located inside a trade stand towards the fire expansion over three groups of compartments, by highlighting the heat transfer, both in small spaces, as well as over large distances. In order to confirm the accuracy of the simulation, the obtained values are compared to the ones from the specialized literature.


Key words: CFD, FDS, simulation, burning, fire

## INTRODUCTION

In order to understand, predict and describe the development of a burning and its' influence on the surrounding environment, there are used a series of scientific methods in whose development are involved both practical experiments and mathematical modelling. The achievement of real scale experiments or even at a proportionally decreased scale, involves substantial costs, as well as a high work volume. Once with the development of the knowledge from this field and with the computers' processing power increase, the prediction of the volume and direction of smoke, of the toxic and flammable gas concentrations and the calculation of pressures and temperatures due to burnings have experienced a significant progress and have decreased the costs to a reasonable level.

Computational modelling of fires in closed spaces comprises two distinct methods: zone type modelling and field type modelling. The first method involves dividing the considered space in two distinct control volumes: an upper volume containing burned and hot gases engaged by the fire, and a lower volume with fresh and cold air content.

Through the second method, the domain is divided in a large number of control volumes, proportional to the dimensions of the objects comprised in the simulation. For these volumes, through the CFD technique (Computational Fluid Dynamic), there are calculated the state equations of the gases and the mass, moment and energy conservation equations. In case of fires, there are used both models for combustion and models for turbulences and thermal radiations, flame propagation, ashes quantity, etc. This method has become not only a tool for the researchers but it also has applicability in the conventional engineering of fires for the optimization of the buildings' safety.

The accuracy of computational simulations depends, for example, on the resolution of the network of volumes, but especially on the limitations and uncertainties of the applications from this relatively new domain.

## SIMULATION OF A FIRE OCCURRED IN A HYPOTHETICAL WHOLESALE WAREHOUSE

The objectives of the virtual simulation of the fire using FDS software aims to estimate the temperatures occurred at the level of a trade stand in which there were stored various goods of more or less combustible content, the development of the fire from the occurrence of the ignition source until the burning of the entire trade stand, and depending on these estimated temperatures, the approximation of the flame front propagation outside the compartment and released heat quantity, hot gases and smoke migration inside the wholesale warehouse [2].

## The development of the fire inside a compartment

In order to initiate the burning inside a compartment, there has been taken into account a supposed ignition source located in a trash bin from nearby a wooden shelf. At a 0.154 m distance from the trash bin there is a cardboard box. On the shelves there are stored textiles and other cardboard boxes.

In the middle of the room there is a chair and a table, both made of wood. The trading stand is made of PVC carpentry (Fig. 1).


Fig. 1 Geometry of the trading stand

| Material/ <br> Substance | Density $\left[\mathrm{kg} / \mathrm{m}^{3}\right]$ | Specific heat $[\mathrm{kJ} /(\mathrm{kg} \cdot \mathrm{~K}]$ | Thermal conductivity $[\mathrm{W} /(\mathrm{m} \cdot \mathrm{~K})]$ | Heat release <br> [kJ/kg] | Heat release per unit area $\left[\mathrm{kW} / \mathrm{m}^{2}\right]$ | Ignition temperature $\left[{ }^{\circ} \mathrm{C}\right]$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Wood | 640 | 2.85 | 0.14 | 17000 | 200 | 240 |
|  |  | $1.29\left(23^{\circ} \mathrm{C}\right)$ | $0.192\left(23^{\circ} \mathrm{C}\right)$ |  |  |  |
| PVC | 1380 | $1.35\left(50^{\circ} \mathrm{C}\right)$ | $0.175\left(50^{\circ} \mathrm{C}\right)$ | 20000 | 312 | 390 |
|  |  | 1.47 (100 $\left.{ }^{\circ} \mathrm{C}\right)$ | $0.147\left(100^{\circ} \mathrm{C}\right)$ |  |  |  |
|  |  | $1.59\left({ }^{\circ} \mathrm{C}\right)$ | $0.134\left({ }^{\circ} \mathrm{C}\right)$ |  |  |  |
| Corrugated cardboard | 250 | 1.336 | 0.05 | 18000 | 150 | 218-246 |
| Plastic | 950 | 1.67 | 0.18 | 31000 | 400 | 349 |
| Textiles | 240 | 1.357 | 0.1 | 17000 | 250 | 220 |

After the discretization of the geometry, there resulted a mesh with 718.008 cells. The trading stand has a 3.75 length, a width of 3.37 m and the height of 3.4 m , resulting in a total volume of 42.97 m 3 . The distances between the shelves vary between 0.5 and 0.7 m , and the distance between the top shelf and the ceiling is 0.96 m .

The atmospheric conditions frame into the normal limits, meaning:

- atmospheric pressure: $1.01325 \times 105 \mathrm{~Pa}$,
- temperature inside the warehouse: $20^{\circ} \mathrm{C}$,
- relative humidity: 40\%.

The characteristics of the materials used are presented in Table 1 [1, 3, 4, 6, 7].

## Ignition source

In the tested scenario, the ignition source initiates the burning process of the plastic bin within a time periods which represents the sum of the pyrolysis time of the solid and the induction time of the gaseous phase.

Once with the flow rate increase, the heat flux per area increases, and the required time for heating and for the pyrolysis of the solid decreases consequently. On the other hand, an increase of the gas rate will result in a decrease of the flux residence time. This phenomenon will delay the onset of the chemical reaction or will even avoid its' initiation. The presence of a driven (piloted) ignition affects the combustion process by local decrease of induction time. The solid heating is the predominant mechanism in the combustion process, with a low rate of the gas flux and with a normal concentration of oxygen. Fig. 2 presents this ignition model, having a very short time of ignition - approx. 7.8 seconds.


In case of a non-driven ignition (Fig. 3), the induction time increases significantly, and flames occur spontaneously as a result of the progressive heating of the combustible material. At the same time of 300 seconds from enabling the ignition source, the maximum temperature reached has been $53^{\circ} \mathrm{C}$ and increasing.


## Fig. 3 Non-driven ignition

## Flame propagation

Flame propagation over solid surfaces may be seen as a continuous process of driven ignition. In order for the flame to propagate, in front of it there must be transferred enough heat from the flame or from other heat sources. Vaporized fuel is then released and convectively transported outside the area, where it mixes with oxidants and forms the flammable mixture in the area from the front of the flame, following to be ignited by it.

## Heat transfer

Based on the 2nd thermodynamic principle, heat transfer has three components:

- transfer by radiation,
- transfer by conduction,
- transfer by convection.

Fig. 4 presents the three phases of the thermal transfer. The $1^{\text {st }}$ phase (Fig. 4a) represents a start moment in heat transfer from the object containing the source, object already in the burning process, towards the neighbouring cardboard box. This transfer is carried out horizontally, and it is based more on radiative transport, phenomenon amplified in the $2^{\text {nd }}$ phase, once with the increase in temperature.


Fig. 4 Heat transfer:
a) heat transfer - 1st phase, b) heat transfer - 2nd phase, c) heat transfer - 3rd phase

c)

d)


Fig. 5 Flame propagation inside the trade stand
The convective transport of heat starts also in the 1st phase, being carried out by hot gases elevation towards the above shelves, pre-heating their material.

In the $2^{\text {nd }}$ phase, the radiative transport of heat brings the temperature of the boxes' surface to the value of nondrive ignition, so to a value at which a spontaneous ignition of the material occurs and implicitly to an increase of the hot gases quantity which rises towards the ceiling of the trading stand (Fig. 4b).

The $3^{\text {rd }}$ phase is characterised by an increase in heat flux towards the shelves from the upper proximity, flux which is achieved through convective transfer based on hot gases elevation (Fig. 4c). The temperature of the shelves reaches and exceeds the ignition threshold, the front of the flame spreading itself in the compartment.

The various phases of fire evolution within a compartment are presented in Fig. 5 ( $a, b, c, d$ ), and the moment when the flame propagates outside the trading stand is presented in Fig. 6.


Fig. 6 Flame propagation outside the trade stand

## Heat release rate

The development of a fire is generally characterised in terms of released heat rate versus time. The released heat rate is not a fundamental property of a fuel and that is why it cannot be calculated from the main properties of the material, being usually determined through tests.

Table 2 highlights the heat release rate, measured in several situations [5].

Table 2
Heat release rate for some materials and situations

| Fuel | Heat release <br> rate (HRR) |
| :--- | :---: |
| A burning cigarette | 5 W |
| A typical light bulb | 60 W |
| A burning candle | 80 W |
| A human being at normal exertion | 100 W |
| A burning wastepaper basket | 100 kW |
| A burning $1 \mathrm{~m}^{2}$ pool of gasoline | 2.5 MW |
| Burning wood pallets, stacked to a <br> height of 3 m | 7 MW |
| Burning polystyrene jars, in $2 \mathrm{~m}^{2}$ cartons <br> 4.9 m high <br> Output from a typical reactor at an NPP | $30-40 \mathrm{MW}$ |

In case of a simulated fire, following the computerized analysis at the level of one arbitrary compartment considered and supposed to hold the ignition source, the heat release rate graph of values has been obtained (Fig. 7).

Comparing the resulted value of almost 5 MW , of the heat release rate from a burnt compartment comprising a small volume of combustible materials with the volume of the entire trading stand which has a 7 MW value (as stated
in Table 2), measured when burning a stack of wooden pallets with a height of 3 meters [5], the computational analysis can be considered to be very close to the real burning.


## Fig. 7 Heat release rate

The result of the analysis is completed with a series of parameters updated during the simulation, the maximum estimative values and the graphic representations of the most significant ones being:

- heat flux - MAX and Heat flux - MEAN,

- convective heat flux - MAX and Convective heat flux - MEAN

- conductivity - MAX and Conductivity - MEAN,


Fire development in the wholesale warehouse
The computational analysis of the fire at this scale ( $90 \times 84 \times 7 \mathrm{~m}$ ) has required the discretisation of the entire geometry (Fig. 8) into a number of 3.628 .800 cells, meshing which has also been split into 6 sub-meshes in order to take advantage of the benefices brought by parallel processing.


Fig. 8 Geometry of the wholesale warehouse


## Fig. 9 Processing time

Processing a calculation cycle involving the entire geometry of the trading platform is extended to 38:44 hours for a simulation period of 600 seconds, which delays a lot the achievement of results close to the real ones and limits the time period for the simulation to be carried out (Fig. 9). The distance between two rows of compartments (width of the access pathways) is 2.8 m . However, Fig. 10a highlights the radiative hear transport over this distance and the increase in temperature of the trade stand located across the source stand, leading to its' ignition (Fig. 10b).


Fig. 10 Radiative heat transport:
a) radiative transport between two trade stands b) Ignition of the stand located at 2.8 m from the fire-place


Fig. 11 Ignition of combustible materials located far from the fire-place:

## a) onvective heat transport, b) expansion of the burning

The heat release rate by the burning of three rows of stands is presented in Fig. 12, hence the estimated value of 330 MW.

Other estimated values are shown in Table 3.


Fig. 12 Heat release rate
Table 3
Estimated values of temperature transport

| Parameter | Value | Measurement unit |
| :--- | :--- | :---: |
| Radiative Loss | $2,25 \times 105$ | kW |
| Conductive Loss | $2,8 \times 105$ | kW |
| Convective Loss | -3750 | kW |

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## CONCLUSIONS

Mathematical modelling of potential high-risk events allows the estimation at a reasonable cost of the risks that may be subject to personnel, buildings and environment. Although it requires a large volume of information for the input data, multiple tests with considerable processing times even for a simple simulation, the prediction of such events may be a barrier between "to happen" or "not to happen". Estimations through mathematical modelling thereby become an amazing resource in the in the occupational health and safety field, environmental protection, design, regardless of the industry branch.

However, due to limitations imposed by the processing power, it is almost impossible to simulate the entire history of a long lasting event, using time and space settings close to real ones, and a sufficiently fine resolution of the meshing in order to obtain accurate results. Yet, a simulation performed over a time period of tens to hundreds of seconds may result in real advantages with immediate applicability.

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